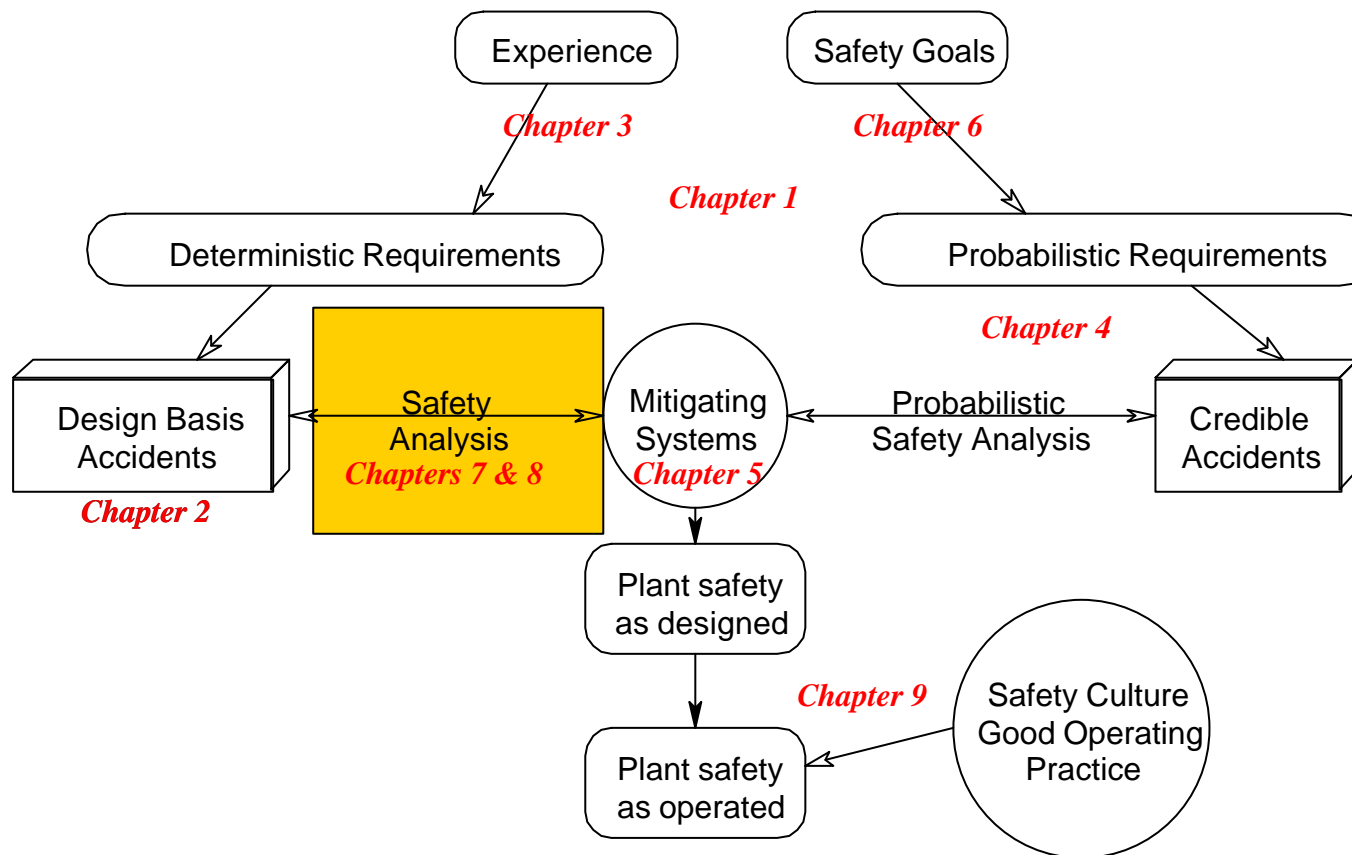


Lecture 11 – Technology of Accident Analysis



Dr. V.G. Snell

Where We Are





Reactor Physics - Revisited

- Recall for a point reactor:

$$\frac{dN_f(t)}{dt} = \frac{k_\infty(r - b)}{l_p} N_f(t) + \sum_{i=1}^6 \lambda_i C_i$$

$$\frac{dC_i}{dt} = \frac{\beta_i N_f(t)}{l_p(1 - r)} - \lambda_i C_i$$



CANDU is Not a Point Reactor

- Flux tilts from movement of adjusters, varying zones, fuelling, xenon
- Flux tilt in accidents from half-core void, insertion of shutoff rods from top
- 3-D diffusion + point kinetics
 - Neutrons are like flow through medium



Continuity Equation - Production

Let $n(\mathbf{r}, t)$ be neutron density at point \mathbf{r} and time t

- assume all at same speed

$$\frac{d}{dt} \int_V n(\mathbf{r}, t) dV = \text{production} - \text{absorption} - \text{leakage}$$

Let $s(\mathbf{r}, t)$ be # of neutrons /vol/time emitted at point \mathbf{r} and time t

$$\text{production} = \int_V s(\mathbf{r}, t) dV$$

Continuity Equation – Absorption

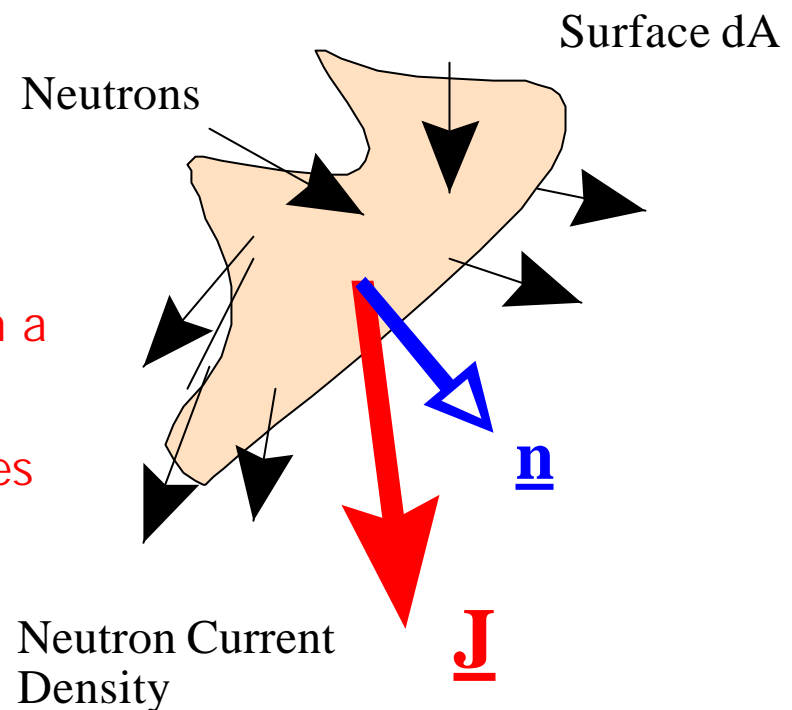
Similarly

$$absorption = \int_V \Sigma_a(\mathbf{r}) f(\mathbf{r}, t) dV$$

$f(\mathbf{r}, t)$ is flux

- Total rate at which neutrons pass through a given area, regardless of orientation
- Useful for describing neutron reaction rates

$\Sigma_a(\mathbf{r})$ is the absorption cross section





Continuity Equation – Leakage

Let $\mathbf{J}(r, t)$ = neutron current density vector

- measures the net flow of neutrons across a unit area in any given direction

Let \mathbf{n} be a unit normal vector pointing outward from the surface \mathbf{A} around \mathbf{V}

$$leakage = \int_A \mathbf{J}(\mathbf{r}, t) \cdot \mathbf{n} dA$$



Continuity Equation

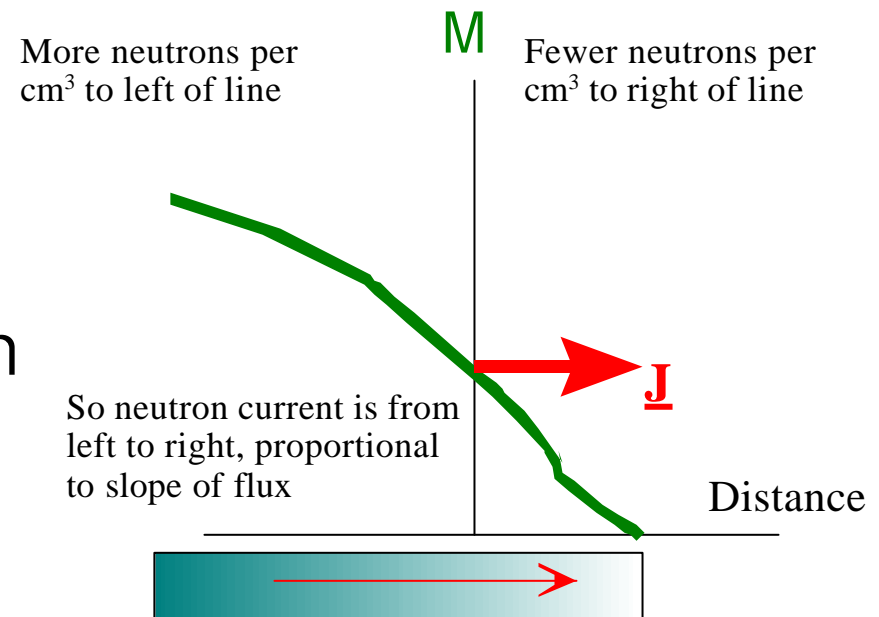
$$\frac{d}{dt} \int_V n(\mathbf{r}, t) dV = \int_V s(\mathbf{r}, t) dV - \int_V \Sigma_a(\mathbf{r}) f(\mathbf{r}, t) dV - \int_V \nabla \cdot \mathbf{J}(\mathbf{r}, t) dV$$

$$\frac{\partial n(\mathbf{r}, t)}{\partial t} = s(\mathbf{r}, t) - \Sigma_a(\mathbf{r}) f(\mathbf{r}, t) - \nabla \cdot \mathbf{J}(\mathbf{r}, t)$$

Fick's Law

- Current density vector \propto negative gradient of the flux
- Proportionality constant is diffusion coefficient, D

$$\mathbf{J} = -D\vec{\nabla} f$$





Neutron Diffusion Equation

- For single energy

$$D\nabla^2 f - \Sigma_a f + s = \frac{1}{v} \frac{\partial f}{\partial t}$$

- Compare heat conduction

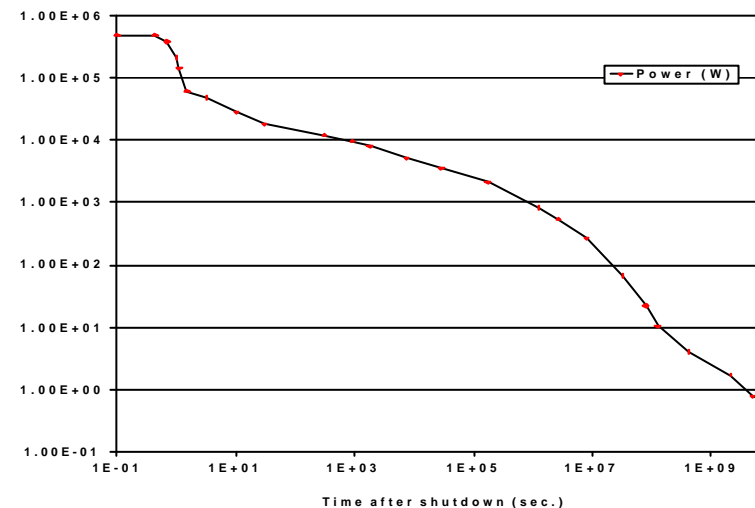
$$\rho c \frac{\partial T}{\partial t} = H + k\nabla^2 T$$

Decay Heat

$$P_d(t) = \sum_i n_i(t)E_i$$

- $P_d(t)$ - power produced by all decaying fission products at time t
- $n_i(t)$ - number of atoms decaying per unit time of fission product i at time t
- E_i - average energy produced by the decay of each atom of fission product i

CANDU Bundle Power after Shutdown





Fuel

- Key safety parameters
 - Fuel temperature
 - Drives fission product transport
 - Potential sheath failure
 - Potential pressure-tube failure
 - Limited effect on physics
 - Fuel sheath integrity
 - Fission product inventory & release

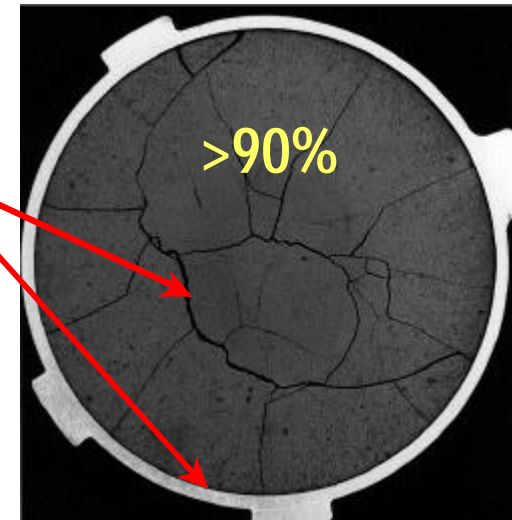
Location of Fission Products

- Fission products formed within fuel grains
- Diffuse
 - Bound inventory – in grains
 - Grain boundary inventory
 - Gap inventory

Fission products move this way with increasing temperature & burnup

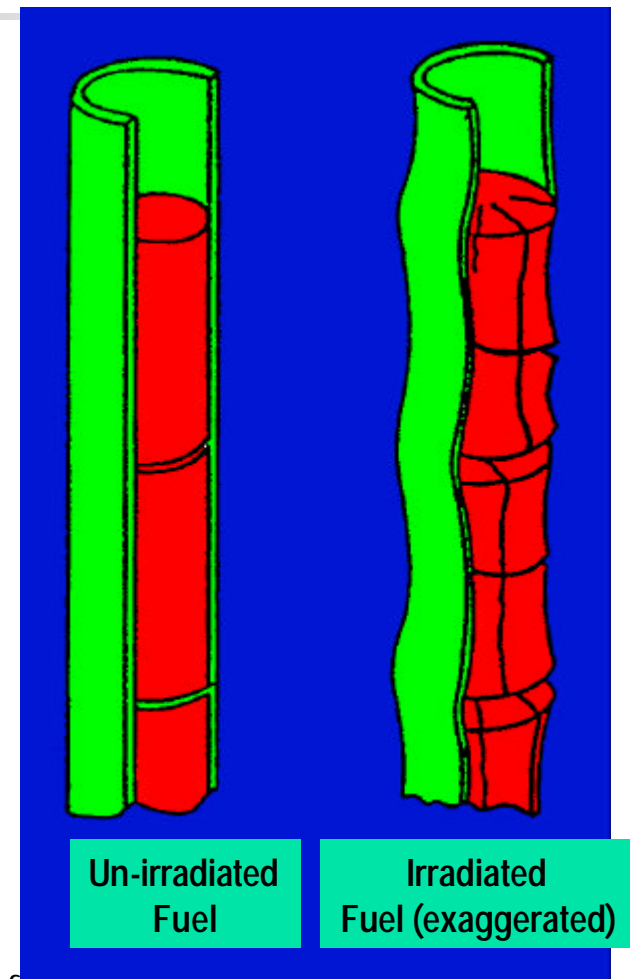
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Behaviour With Irradiation

- Cracking
- Swelling
- Dishing/ridging
- Gas pressure increase
- Pellet-clad interaction





Fuel Heat Conduction

One dimension $Q = -kA \frac{dT}{dx}$

Three dimensions

(Rate of change of internal energy) =
(rate of energy release) - (rate of energy loss from conduction)

$$rc \frac{\partial T}{\partial t} = H + k \nabla^2 T$$

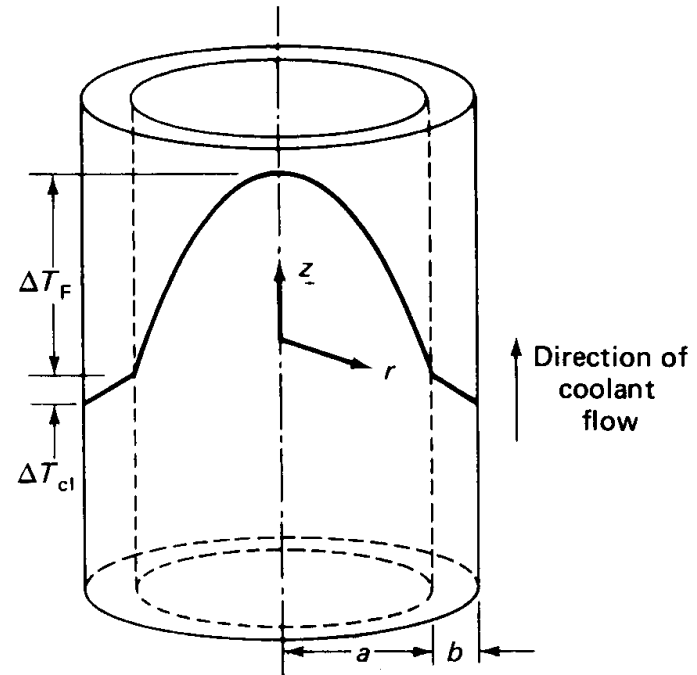
Cylindrical Fuel Pin

Steady state heat conduction

$$\frac{d^2 T}{dr^2} + \frac{1}{r} \frac{dT}{dr} = -\frac{H}{k_F}$$

Integrate (let $u = r \frac{dT}{dr}$)

$$T(r) = T(0) - \frac{Hr^2}{4k_F}$$





Sheath and Gap

Apply same equation to sheath

$$\Delta T_S = T_{Si} - T_{So} = \frac{Ha^2 \log[(a+b)/a]}{2k_s}$$

And gap

$$q = h_g (T_F - T_{Si})$$

And coolant

$$q = h(T_{So} - T_C)$$



Sheath-to-Coolant ΔT

Steady State –

All heat produced in fuel is transferred to coolant, so for length l :

$$q = \frac{Hp a^2 \ell}{2p(a+b)\ell}$$

$$T_C - T_{So} = \frac{Ha^2}{2h(a+b)}$$

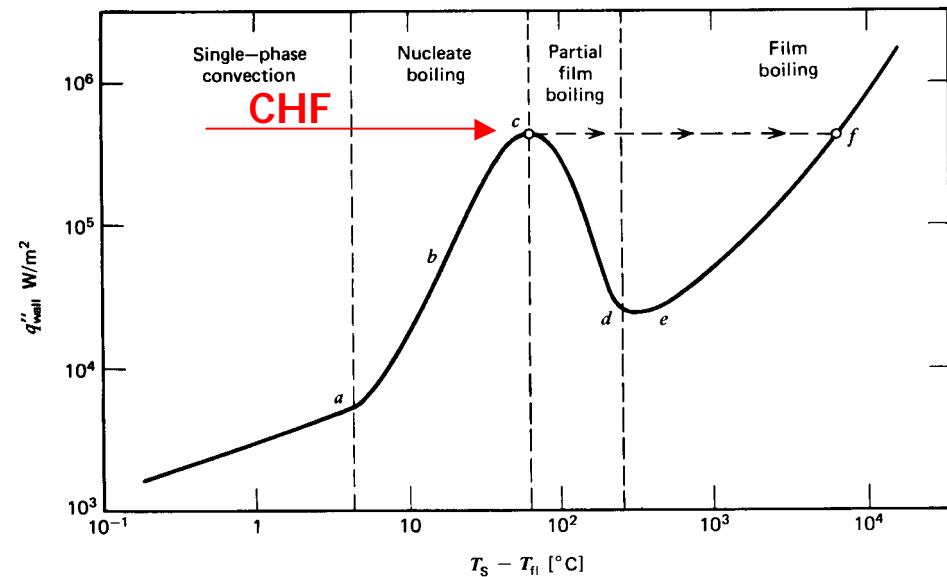


Safety Importance

Characteristic	UO ₂ Fuel	Metal Fuel
Thermal Conductivity	Low	High
Melting Point	High	Low
Heat Capacity	High	Low

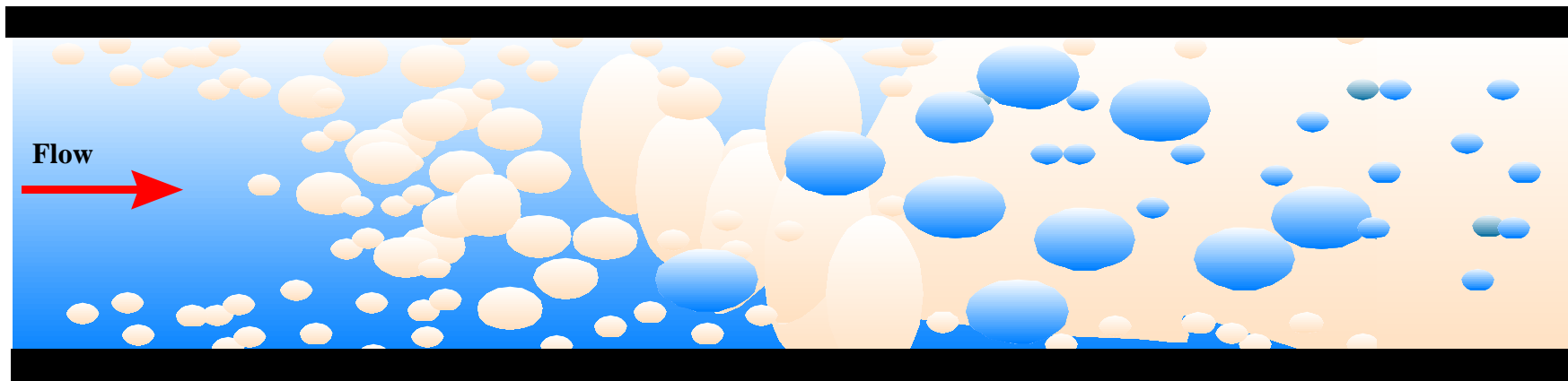
Dryout

- Sudden drop in sheath-to-coolant heat transfer when “Critical Heat Flux” is reached
- Temperature jump strongly dependent on subcooling



Heat flux versus temperature difference for pool-boiling heat transfer.

Flow Regimes in Horizontal Heated Channel (High Flow)



**Subcooled
Boiling**

**Saturated
Nucleate
Boiling**

**Forced
Convective
Evaporation**

**Film
Boiling**

**Forced
Convection
to Vapour**

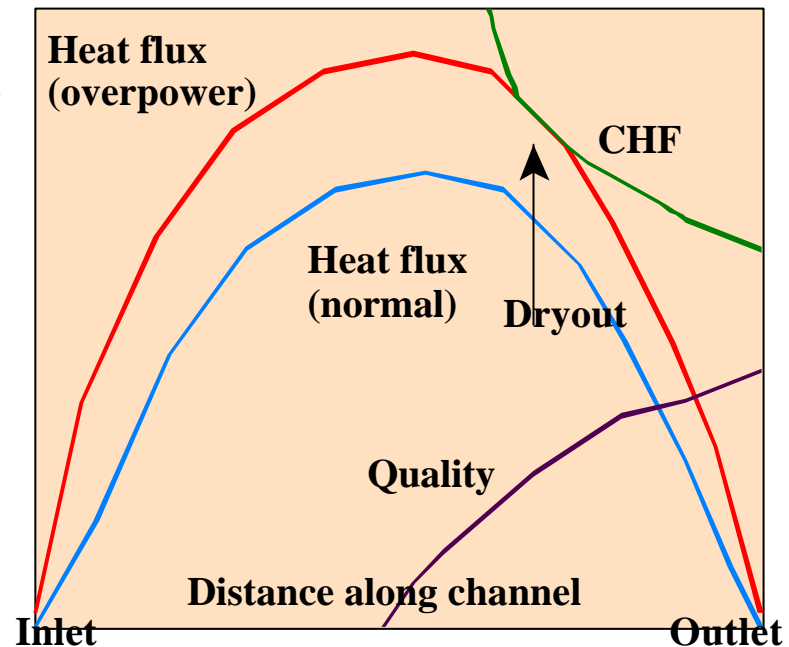


Progressive Effects of Overpower

- Dryout
- Sheath temperature rise
- Zircaloy annealing
- Oxidation embrittlement of sheath
- Braze melting and attack on Zircaloy
- Zirconium-water reaction (exothermic)
- Bundle collapse
- Sheath melting
- Fuel melting (extremely unlikely)
- Pressure tube balloon or burst
- Heat transfer to moderator

CHF in a CANDU Channel

- CHF determined experimentally
 - no reliable theory for needed accuracy
- Local flux shape means dryout is not at the end
- How can we change the flux shape to improve margins?





Gas Pressure

- Driving force for sheath strain in accidents
- Affects sheath liftoff
 - Therefore fuel-to-sheath heat transfer
 - Therefore fuel temperature
- Model via ideal gas law



Strain

- Relevant to large LOCA
- Transverse strain:

$$S = \frac{Pr}{w}$$

- P is the pressure differential across the tube
- r is the tube radius
- w is the tube thickness



Strain Rate Equations

$$\dot{\epsilon} = \frac{d\epsilon}{dt} = A s^n e^{-k/T} + B s^m e^{-\ell/T}$$

All parameters determined from experiment

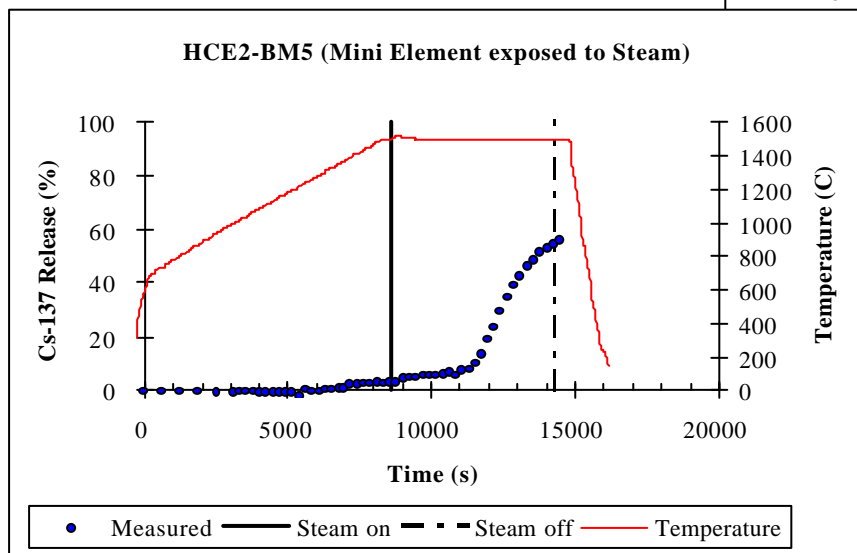
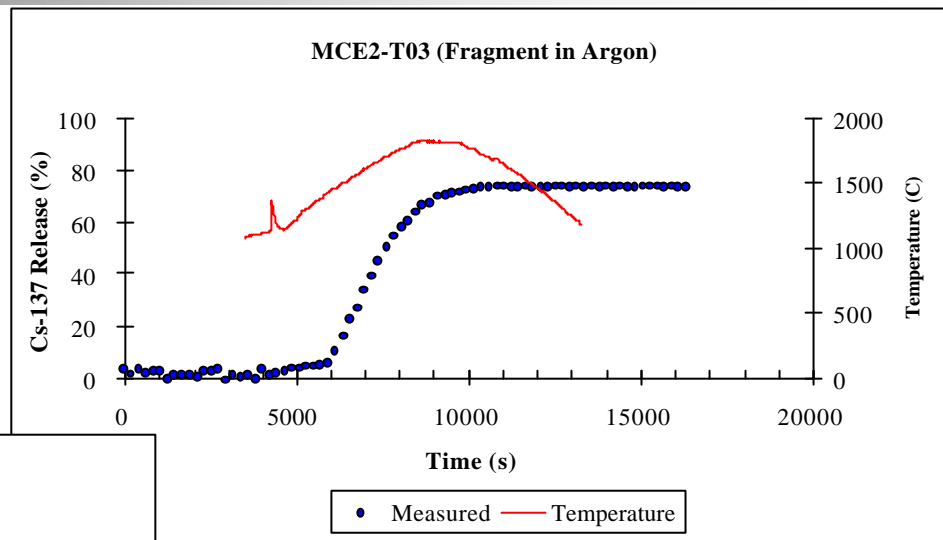
Sheath may fail by ballooning if $\epsilon > 5\%$

What is fission product release?

- fraction of gap inventory
- small % of grain-boundary & bound inventory
 - at high temperatures only

Fission Product Release

Release is a strong function of temperature, atmosphere



Technology of